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PHOTO CONDUCTIVITY OF LIQUID TETRAMETHYLSILANE

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W. F. Schmidt and W. Döldissen

Hahn-Meitner-Institut für Kernforschung Berlin GmbH, Bereich Strahlenchemie D-1000 Berlin 39

U. Hahn and E. E. Koch

Deutsches Elektronen-Synchrotron DESY, Hamburg

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W.F. Schmidt and W. Döldissen
Hahn-Meitner-Institut für Kernforschung Berlin GmbH,
Bereich Strahlenchemie, D-1000 Berlin 39

and

U. Hahn and E.E. Koch

Deutsches Elektronen Synchrotron DESY
D-2000 Hamburg 52

Abstract

Intrinsic photo conductivity of liquid tetramethylsilane was observed in the wave length region from 1100 to 1500 Å. The photo ionization threshold E_{I_i} has been determined to be $E_{I_i} = 8.1$ eV.

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Very few investigations have been carried out so far on the direct photo ionization of non-polar liquids. Experiments by Vermeil et al. 1 detected an increase of conductivity in some liquid hydrocarbons when subjected to the light of rare gas resonance lamps. Recently similar experiments were performed by Casanovas and co-workers at the University of Toulouse², 3. It was estimated that the ionization threshold energy of the molecules in the gas phase is reduced for the different liquids by 0.5 eV to 1 eV upon condensation.

Direct measurements of the ionization threshold energy for liquid xenon were reported by Roberts and Wilson¹, Asaf and Steinberger⁵ and Spear and LeComber⁶. While Roberts and Wilson found a strong increase of the photo current with photon energy above 8.9 eV, Asaf and Steinberger determined the threshold energy to 9.2 eV, which agrees with the value reported by Spear and LeComber.

The photo ionization threshold in the liquid phase \mathbf{E}_{L} should be given by

$$E_{L} = E_{G} + V_{O} + P_{+} \tag{1}$$

with \mathbf{E}_{G} , gas phase energy; \mathbf{P}_{+} , polarization energy of the positive ion and \mathbf{V}_{o} , electron affinity of the liquid.

The electron affinity or the energy of the electronic conduction level has been determined for many hydrocarbons and liquefied rare gases by means of the photoelectric effect on a metal electrode $^{7-9}$. The polarization energy of the positive ion can then be obtained if $\rm E_L$ is measured. On the other hand, $\rm P_+$ can be estimated from Born's equation 16 which yields for an ion of radius r

$$P = -\frac{e^2}{2r} (1 - \frac{1}{t})$$
 (2)

with ϵ the relative dielectric constant of the liquid and ϵ the elementary charge.

Here we wish to report the results of photoconductivity experiments in liquid tetramethylsilane ((CH3)2Si) at 22°C. The liquid was contained in a parallel plate conductivity cell with 2 mm plate separation. One electrode consisted of a thin gold layer (thickness approximately 150 Å) evaporated on to a LiF plate which served as the entrance window for the radiation. The counter electrode was made of gold plated brass, the cell body was of stainless steel. Synchrotron radiation from the storage ring DORIS was monochromatized by a high intensity monochromator 11 . The resolution was adjusted to approximately 0.1 eV. With DORIS operating in the single bunch mode (10 - 20 mA current of circulating electrons) an intensity of about 10 10 photons/sec was available at the entrance window of the conductivity cell. In the multi bunch mode (200 - 400 mA current) the intensity would be increased at least by a factor of 10. DC-voltages up to 3 kV were applied across the liquid gap and the photocurrents were measured with a Keithley Mod 602 electrometer. At a particular voltage the wavelength of the radiation was scanned from 1100 ${\hat {\bf A}}$ to 1600 ${\hat {\bf A}}$ and the photo current was recorded. Depending on the applied voltage the photo currents exceeded the dark durrents by a factor of up to 4. The dark currents were subtracted from the currents under illumination in order to obtain the photo currents. Figure 1a shows a typical photo conductivity spectrum obtained. Data were recorded for different polarities and voltages. The spectral dependence of the incident intensity, the transmission of the cell window and the gold alectrode was determined by observing the luminescence intensity of sodium

salicylate placed behind the gold film (fig.1b). In fig. 1c the corrected photo current as a function of wavelength is shown. The current begins to rise below 1500 $\mathring{\rm A}$ and a second steep rise is observed below 1500 $\mathring{\rm A}$.

Since the intensity of the synchrotron radiation decreased with time photo conductivity spectra taken at different times were normalized according to the magnitude of the electron current in the storage ring.

In order to determine the threshold $\mathbf{E}_{\mathbf{L}}$, the photo conductivity c in arbitrary units as a function of photon energy was plotted and a power function was fitted to the points. A surprisingly good fit was obtained with

$$\sigma \propto (E - E_{\rm t})^{3/2} \tag{3}$$

and $\rm E_L$ = 8.1 eV (fig. 2). For the gas phase ionization energy Jonas et al. ¹² reported $\rm E_G$ = 9.79 eV. $\rm V_O$ was determined by Holroyd et al. and the mean of the values reported by them is $\rm V_O$ = -0.57 eV^{7,8,13}. From eq.1 $\rm P_+$ = -1.12 eV is obtained which yields an ionic radius of r = 2.95 $\rm A^{\circ}$ comparable to the hard core radius of the molecule.

The second threshold observed in liquid tetramethylsilane at around 10.75 eV may be compared with the rise in the photo electron spectrum from the gas at 12.2 eV. The difference is 1.45 eV as compared to 1.65 eV for the first threshold. This discrepancy may be due to the fact that the second threshold in the liquid could not be determined from the data in the same way as the first threshold. The photo current due to the second ionization process still contains a considerable contribution due to the first ionization, which impedes the exact determination of this threshold.

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Figure Legends

- Fig.! a) Observed photo current as a function of wave length for líquid Tetramethylsilane; applied voltage = 1300 V, $i_{\bar{D}}$ = 5.5 pA, maximum photo current i_{max} = 19.5 pA.
 - b) Efficiency of the monochromator and transmission of LiF-window and gold layer measured as the photo current of a photomultiplier produced by the luminescence of sodium salicylate.
 - c) Photo current per incident photon, corrected data of fig. la (see text).
- Fig. 2 Computer averaged dependence of photo conductivity on photon energy for liquid Tetramethylsilane (this plot contains 1) photo conductivitiy spectra for different voltages and intensities); The points have been calculated according to eq.3. In the insert a schematic energy level diagrams for gaseous and liquid Tetramethylsilane is shown.

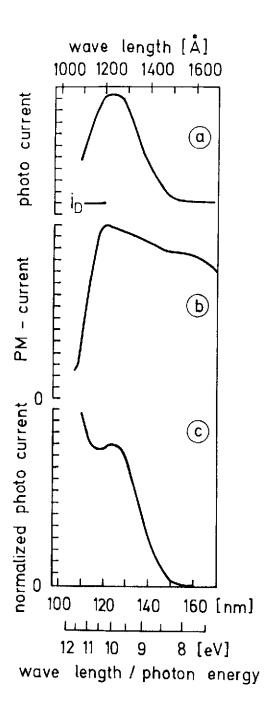


Fig. 1

Show conductivity [arb. units]

Conductivity [ar

Fig. 2