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by

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Abstract

The X-ray L absorption spectra were measured for the atomic rare earths Ce, Sm, Gd and Er. From the energy positions of well resolved absorption lines at threshold in comparison to the corresponding solid state spectra the atom-to-metal shifts of the core level binding energies have been determined. For Sm and Er which undergo a valence change on formation of the solid shifts differ by more than 6 eV from those of Ce and Gd which have the same configuration in the atom and the solid.

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Introduction

of core level binding energies of atoms in different chemical Shifts environments reflect changes in valence orbitals and in the valence configuration and differences in the relaxation processes. The fingerprint character of these shifts explains the considerable experimental and theoretical effort devoted to the investigation of this subject 1,2. Especially the atom-to-metal shifts of the rare earth elements recently have attracted much interest. This interest has been stimulated by the valence fluctuations detected in pure rare earths and rare earth comnounds^{3,4,5}. One of the major problems encountered in this field is the lack of experimental data on free metal atoms in the X-ray region. There are almost no X-ray absorption, X-ray emission and X-ray photoemission data on free metal atoms, from which core-level binding energies, transition probabilities, core-hole lifetimes and indications of the importance of many electron effects can be obtained. In addition to being important for atomic physics.all these values are necessary for a thorough assessment of the influence of the chemical environment, Oscillations in the differential optical oscillator strength or more important, multiple electron excitations can cause structures in the absorption spectra above inner shell thresholds. These effects have to be considered in the analysis of the X-ray absorption near edge structure (XANES)⁸ and the extended X-ray absorption fine structure (EXAFS)⁹ of solids.

The high temperatures required for the preparation of free rare earth atoms and their low absorption cross sections render these experiments extremely difficult. Since a lot of basic information can be obtained from absorption spectra, we measured the L-absorption spectra of a representative set of atomic rare earth elements, Ce, Sm, Gd and Er. Experiment

The measurments were done at the X-ray spectrometer ROEMO at the Hamburger Synchrotron Radiation Laboratory HASYLAB. We have used a non dispersive Si double crystal monochromator optimized for small harmonic content¹⁰. To balance high energy resolution and higher order contamination, (111) Bragg reflection was used for Ce and Sm and (220) Bragg reflection for Gd and Er. The atomic metals were maintained inside a resistence heated tubular furnace¹ mounted behind the exit slit of the monochromator. The metal vapours were contained in a Ta tube except for Er where a W tube was used. The temperatures ranged from 1320 K to 2500 K namely 1320 + 50 K for Sm, 2020 + 50 K for Er, 2250 + 50 K for Gd and 2500 + 50 K for Ce, which according to vapour pressure data¹² correspond to pressures between 2 mbar and 16 mbar. The length of the vapour column was approximately 50 cm. Be windows (400 µm thick) separated the vapour region from the surrounding air. A Ne buffer gas (20 mbar to 30 mbar) retarded the escape of the metal atoms from the hot zone of the furnace and prevented them from reaching the Be-windows. Charging the furnace with a sample of ~ 50 g made possible stable operation for \sim 4-6 hours. For the exact determination of the atomto-metal shifts, the spectra of the rare earth metals were recorded simultaneously with the vapour spectra by using three ionisation chambers in a row. The metal sample was placed between the first and the second and the vapour between the second and third chamber. In some cases metal runs were taken immediately before or after each vapour measurement.

Experimental results and discussion

The L₃ absorption spectra of atomic and metallic Ce, Sm, Gd and Er are presented in Fig. 1. Since the L₂ spectra are very similar to the L₃ spectra they are not shown. The prominent white lines at the threshold of the atomic L₂ and L₃ absorption are attributed to $2p_{1/2,3/2} + 5d$ transitions. As demonstrated in Fig. 1 these lines can be very well fitted by a single Lorentzian. Autoionisation, which could cause an asymmetric line shape 13. does not contribute significantly to the decay of the 2p core hole. The onset of transitions to higher excited states manifests itself on the high energy side in the deviation of the experimental curve from the Lorentzian. A break in the atomic absorption spectrum , which is clearly to be seen $\sim 3 \ eV$ above the center of the white line, is most likely due to 2p + 6d transitions. On the formation of the solid the white lines are shifted and broadened by the interaction with the neighbouring atoms. Materlik et al. 4 demonstrated that the white line and the near edge structure can be well described by a single-particle band structure calculation. For Ce and Gd which have the same $4f^{n}$ configuration (Ce $4f^{1}$, Gd $4f^{7}$) in the atom and in the solid the centers of the white lines almost coincide (1-2 eV shift). In contrast to this, for Sm and Er, which undergo a change of the 4fⁿ configuration on solidification (Sm $4f^6 + 4f^5$; Er $4f^{12} + 4f^{11}$) the peak of the white line is shifted towards higher energies by about 8 eV.

The L_{j} spectra of the atomic and metallic samples are given in Fig. 2. The atomic spectra display a clearly discernible line at threshold due to 2s \rightarrow 6p transitions. The strength of this line is comparable to the strength of the 2s \rightarrow 6p continuum transitions. Therefore the line is harder to disentangle from the transitions to higher excited states than the 2p \rightarrow 5d absorption lines discussed above. In the solid the hybridisation of the outer s , p and d symmetric orbitals changes the spectra at threshold completely. The metal spectra show a steep rise at threshold followed by a further increase of the cross section towards the maximum located approximately 20 eV above the onset. The experimental spectrum is well reproduced by one-electron band structure cal-

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culations¹⁴. The shoulder at threshold and the dominant maximum at higher energies reflect the structure of the p-projected density of conduction band states. Again the atoms undergoing a valence change (Sm. Er) on solidification and those which do not (Ce, Gd) are easily distinguished. For Ce and Gd the steep rise at threshold of the metal spectra almost coincides with the rise of the atomic cross section whereas for Sm and Er they are separated by more than 5 eV.

From the area under the Lorentzian lines the ratio of the cross sections $\sigma(2p_{3/2} + 5d) / \sigma(2p_{1/2} + 5d)$ has been determined to be 2.1 ± 0.1. Within the error limits this ratio agrees with the statistical ratio of 2 and indicates only small differences between the respective matrix elements. The cross section $\sigma(2s + 6p)$, determined in an analogous way is approximately a factor of 6 smaller than $\sigma(2p_{3/2} + 5d)$.

From the comparison of the experimental and the calculated spectra¹⁴ the metal core level binding energies referred to the Fermi level $E_{2s,F}^{M}$, $E_{2p,F}^{M}$ have been inferred¹⁵. The positions of the maxima of the atomic absorption lines are used to determine the $2p \neq 5d$ and $2s \neq 6p$ excitation energies $E_{2p \neq 5d}^{A}$ and $E_{2s \neq 6p}^{A}$. The energy differences

$$\Delta \mathbf{E} = \mathbf{E}_{\mathbf{ex}}^{\mathbf{A}} - \mathbf{E}_{\mathbf{cF}}^{\mathbf{M}}$$

between the atomic excitation energies E_{ex}^{A} and the metal core level binding energies E_{cF}^{M} are listed in Table 1. Valence changes on solidification can be well identified with the help of these ΔE values. In order to compare our energy shift ΔE with the calculated core-level binding energy shifts

$$\delta \mathbf{E} = \mathbf{E}_{\mathbf{c}}^{\mathbf{A}} - \mathbf{E}_{\mathbf{c}F}^{\mathbf{M}} = \mathbf{E}_{\mathbf{e}\mathbf{x}}^{\mathbf{A}} + \mathbf{E}_{\mathbf{c}}^{\mathbf{*}\mathbf{A}} - \mathbf{E}_{\mathbf{c}F}^{\mathbf{M}}$$

we have to correct for the atomic binding energy of a 5d (E_{5d}^{*A}) or 6p (E_{6p}^{*A}) electron in the presence of a 2p or 2s hole respectively. Since the 2s and 2p ionisation energies cannot be directly read off from the spectra, values for the binding energies E_{5d}^{*A} were used from atomic relativistic Hartree-Fock calculations ¹⁶ to determine δE_{L_2} and δE_{L_3} . For δE_{L_1} no such values were available so that we had recourse to the Z+1 model. Therefore in table 1 approximate values for E_{5d}^{*A} and E_{6p}^{*A} are listed also, which have been obtained from tabulated energy levels of the Z+1 atoms ^{17,18,19}. As seen, values obtained from either method show very good agreement. In this context we want to remind the reader that the interaction of the atomic valence electrons results in a multiplet splitting of 0.5 eV to 1.0 eV ^{17,18,19} which can modify the tabulated shifts correspondingly.

For each element the differences between the δE_{L_1} , δE_{L_2} and δE_{L_3} shifts are within the error limits. In view of the discussion above, the atom-to-metal core-level-independent binding energy shifts calculated by Johanssen and Martensson² are in good agreement with the experimental values. The values reported from the ab-initio calculation by Herbst¹ also agree reasonably well with our measurements and show, even at these deep core states, only small differences. Both experimental and theoretical values show clearly the strong influence of the atom-to-metal values change.

Acknowledgement

We like to thank Dr. J.F. Herbst for making his calculated atomic binding energies $E_c^{\#\Lambda}$ available to us.

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<u>Table 1</u>: Experimental energy separation $\Delta E_{L} = E_{ex}^{A} - E_{cF}^{M}$ and experimental δE_{L} and theoretical^{1,2} δE^{calc} atom-to-metal core-level binding energy shifts. The atomic 5d and 6p binding energies E_{5d}^{*A} and E_{6p}^{*A} are given in the sequence: reference 16 / Z+1 model. $\delta E_{L_{2,3}}$ and $\delta E_{L_{1}}$ were calculated with values from reference 16 and from the Z+1 model respectively. Note that the tabulated error only includes the uncertainty to

determine E_{ex}^{A} and E_{o}^{m} from the experimental data.

	Ce	Sm	Gđ	Er
ΔE _{L3} (eV)	2.6 <u>+</u> 0.3	-3.8 ± 0.3	2.8 <u>+</u> 0.3	-3.7 <u>+</u> 0.3
E <mark>∜A</mark> (eV) 5d (eV)	7.6 / 7.4	6.3 / 6.7	7.6 / -	5.2 / 5.7
^{δE} L3 (eV)	10.2 <u>+</u> 0.3	2.5 <u>+</u> 0.3	10.4 <u>+</u> 0.5	1,5 ± 0,3
∆E _{L2} (eV)	2.6 <u>+</u> 0.3	-3.9 <u>+</u> 0.3	2.9 <u>+</u> 0.3	-3.4 ± 0.3
E <mark>#A</mark> (eV)	7.6 / 7.4	6.3 / 6.7	7.6 / -	5.2 / 5.7
δE _{L2} (eV)	10,2 <u>+</u> 0,3	2.4 <u>+</u> 0.3	10.5 <u>+</u> 0.3	1.8 + 0.3
∆E _{Ll} (eV)	5.3 <u>+</u> 0.6	-2.5 <u>+</u> 0.5	4.7 <u>+</u> 0.4	-3.6 <u>+</u> 0.7
E <mark>#A</mark> (eV)	- / -	- / 5.0	- / -	- / 4.9
δE _{Ll} (eV)		2.5 <u>+</u> 0.5		1.3 ± 0.7
δE ^{calc} ref.1 (4	eV) 9.3	0.4	8,4	0
δE ^{calc} ref.2 (e	≥V) 9.7	1.8	9.1	1.5

Figure Captions

- <u>Fig. 1</u> L₃ absorption spectra of atomic (dotted line) and metallic (solid line) Ce, Sm, Gd and Er. A background, determined by extrapolating the absorption spectrum below the 2p threshold towards higher energies by a straight line, has been subtracted. The positions of the Fermi levels E_F^M , the atomic 5d excitation energies E^A and the ionisation energies E_{5d}^{*A} are indicated. The dashed line gives a Lorentzian absorption profile. The edge heights are normalized. The inflection point E_O^m of the absorption edge of the solid state spectrum is chosen as origin of the energy scale.
- Fig. 2 L_1 absorption spectra of atomic (dotted line) and metallic
(solid line) Ce, Sm, Cd and Er. A background, determined
by extrapolating the absorption spectrum below the 2s
threshold towards higher energies by a straight line,
has been subtracted. The positions of the Fermi levels E_F^M ,
the atomic 6p excitation energies E^A and the ionisation
energies $E_{6p}^{\bullet A}$ are indicated. The edge heights are normalized.
The inflection point E_0^m of the absorption edge of the solid
state spectrum is chosen as origin of the energy scale.



