

Effects of oxygen variation on the improved structural stability, electronic and optical properties of ZnTeO compounds

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Abstract

Crystalline ZnTeO thin films are promising materials for next generation photovoltaics. However, their structural stability and optical nonlinearity potential in bulk form have not been reported. Here, structural, electronic and optical properties of ZnTeO composites have been thoroughly studied using genetic algorithm and density functional theory (DFT). Energetically, mechanically and dynamically stable O-rich phases, namely Zn₂Te₂O₆ and ZnTeO₄, were obtained. Ground-state properties such as lattice constants and simulated XRD were analyzed and compared to the experimental literature wherever possible. With a G_0W_0 corrected band gap, these semiconducting phases display several desirable features, namely, Jahn–Teller distorted cations, hardness and shear anisotropy-induced optical nonlinearity that increase monotonically as O concentration elevates. Such trends appear to be consistent with that seen in the experimental study of ZnTeO thin film. It is observed that Zn-*d*, Te-*p* and O-*p* states have immense influence towards the electronic properties of these structures. Both phases exhibit steep elevation of absorption throughout the ultraviolet (UV) range, hitting peak value of $\sim 5.0 \times 10^5 \text{ cm}^{-1}$. Of particular interest, the non-centrosymmetric ZnTeO₄ has second harmonic generation coefficients (9.84 pm V⁻¹ and 2.33 pm V⁻¹ at static limit) greater than borates crystal and large birefringence that exceeds 0.08 in deep UV region, thus highlighting its potential pedigree as new optical materials in UV range.

Keywords: density functional theory (DFT), electronic properties, linear and nonlinear optics, second harmonic generation (SHG)

 Supplementary material for this article is available [online](#)

(Some figures may appear in colour only in the online journal)

1. Introduction

Most semiconductors are comprised of elements that are comparatively well-matched in terms of atom dimensions, ionicity, and electronegativity. Semiconducting materials in this category have energy band gap that can be exactly estimated using a linear interpolation of the end point values, as described by

the virtual crystal approximation (VCA) [1]. The VCA model can be altered with an inclusion of quadratic bowing parameter term [2, 3] to suit the application in less well-matched semiconductors. Still, there is a class of semiconductors known as highly mismatched alloys (HMAs), which contain elements with very diverse characteristics. These combinations cannot be accurately expressed using the VCA model [4].

HMAs have over the years emerged as key materials for the application of deep ultraviolet optoelectronics, photovoltaic, harsh environment applications and high-power laser devices [5, 6], mainly due to their ultrawide band gap characteristic and distinctive refractive index of 1.4–2.8 within visible radiation that elevates for shorter wavelengths. To create an HMA, host anions (electron-negative metallic anions) are substituted with mismatched anions (isovalent electronegative metallic anions), thus introducing an intermediate band energy level between the valance and conduction band. The composition of mismatched anions will determine the degree of the band structure, or energy gap in short. Initially HMA can only be created through III–V alloys. However, recent advances in thin film technology has made II–VI semiconductors a viable option for the synthesis of HMA [7]. Currently, two most studied HMAs are GaAs doped with N and ZnTe doped with O [8]. These two semiconductors have manifest absorption of sub-bandgap photons, justifying the existence of intermediate band energy level. In II–VI based HMAs, ZnTeO and ZnSeO are of the interest. In the studies of ZnTeO [9–16] and ZnSeO [17, 18], small portion of Te (Se) is replaced with O in dilute oxides, otherwise small fraction of O is replaced with Te (Se) [18–21] in dilute tellurides (selenide) forms.

Interest on ZnTeO arises when it was discovered that ZnTeO-based devices show higher efficiency and photocurrent as compared to ZnTe-based device itself [22]. However, detailed study of their optical nonlinearity potential is extremely lacking. Our literature search also indicates that only one publication on *ab initio* calculations of bulk ZnTeO has been produced so far. Zerroug *et al* [22] used full potential linearized augmented plane wave (FP-LAPW) method as employed in Wien2K to investigate the structural and optical properties of ZnTeO alloys while varying the concentration of oxygen. The calculated static and high-frequency dielectric constants for ZnO and ZnTe end blocks are found to agree reasonably well with the experimental findings. The calculations, nevertheless, were carried out based on the assumption that ZnTeO always maintain the zincblende structure. It must be stressed that local minimum and space group for intermediate oxide ZnTe phases, in premature form, can vary from the pre-conjectured cubic structure. As such, potentially important structures of other symmetries might be missed during the calculation. In our work, we performed a thorough search for whole composition range of oxygen gas within ZnTe–O₂ system, mimicking the influx of O₂ onto ZnTe substance, and compared the formation enthalpy of each fit structure. Being able to characterize efficient 3D optical materials has always been an elusive objective but undeniably important, mainly because of their ability to restrain and regulate EM waves in 3D space [23, 24]. Therefore, findings obtained in our work would conveniently allow researchers or manufacturers to extract instructive data, needed for growing bulk ZnTeO crystals of superior quality by the crystal growth optimization method.

2. Computational method

To identify all probable stable structures in the binary ZnTe–O₂ system, USPEX [25, 26] bound extensive evolutionary calculations that consist of 4–8 atoms, 6–12 atoms, 8–16 atoms and

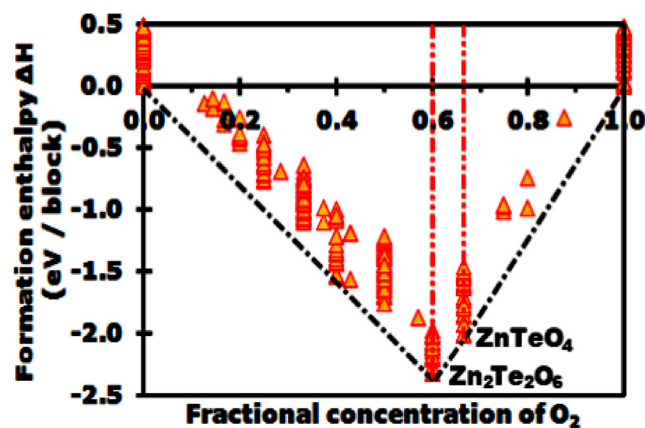


Figure 1. Convex hull for ZnTe–O₂. The lowest free energy of O₂ gas is -5.5953 eV/atom.

10–20 atoms sample sizes were performed under atmospheric pressure. Default atom dependent hard constraints such as minimum distance between atoms and minimum volume per block were accessed. In the first generation, 60 random structures with combined number of atoms per unit cell that satisfy the lower-upper limits of sample range and constraints were created. Each structure underwent five relaxation intervals, starting from unrefined to strict conditions, within the framework of density functional theory (DFT) [27, 28] as provided by the Vienna *ab initio* simulation package (VASP) [29, 30]. Reciprocal space resolution of Monkhorst–Pack *k*-meshes was reduced gradually from $2\pi \times 0.15 \text{ \AA}^{-1}$ to $2\pi \times 0.07 \text{ \AA}^{-1}$. Methfessel and Paxton scheme with smearing width starting at 0.10 eV and reducing to 0.03 eV was preferred for the relaxation stages.

We used pseudopotentials with twelve and six apiece valence electrons for Zn ($3d^{10}4s^2$), Te ($5s^25p^4$) and O ($2s^22p^4$), respectively. Enthalpy or DFT computed Gibbs free energy of locally optimized structure was used as fitness, and 60% of lowest-enthalpy structures participated in new structures formation via crossover and mutations techniques, namely the heredity (50%), random (20%), transmutation (10%), softmutation (10%) and lattice mutation (10%) operators. Addition of randomly generated compositions is imperative as it provides a diversity of structures for every generation. Softmutation refers to the induction of structural change by dislodging atoms along the softest mode eigenvectors. In lattice mutation, lattice vectors of the parent structure are changed by applying a symmetric strain matrix. This operator helps to minimize events of untimely convergence towards a particular lattice. Transmutation turns randomly chosen atoms within a parent structure into other chemical species present in it. Each newly formed offspring was again compared to preset hard restrictions, relaxed and moved into the next generation if survived. Bad candidates with quality aspect determined by the enthalpy value were eliminated. The above cycle was repeated until generation number 50. At this stage, selected low enthalpy individuals have been fully relaxed.

Selected stable and metastable phases were reoptimized with increased precision using a fully automatic $4 \times 4 \times 4$ *k*-point grid Monkhorst–Pack scheme in a static self-consistent run,

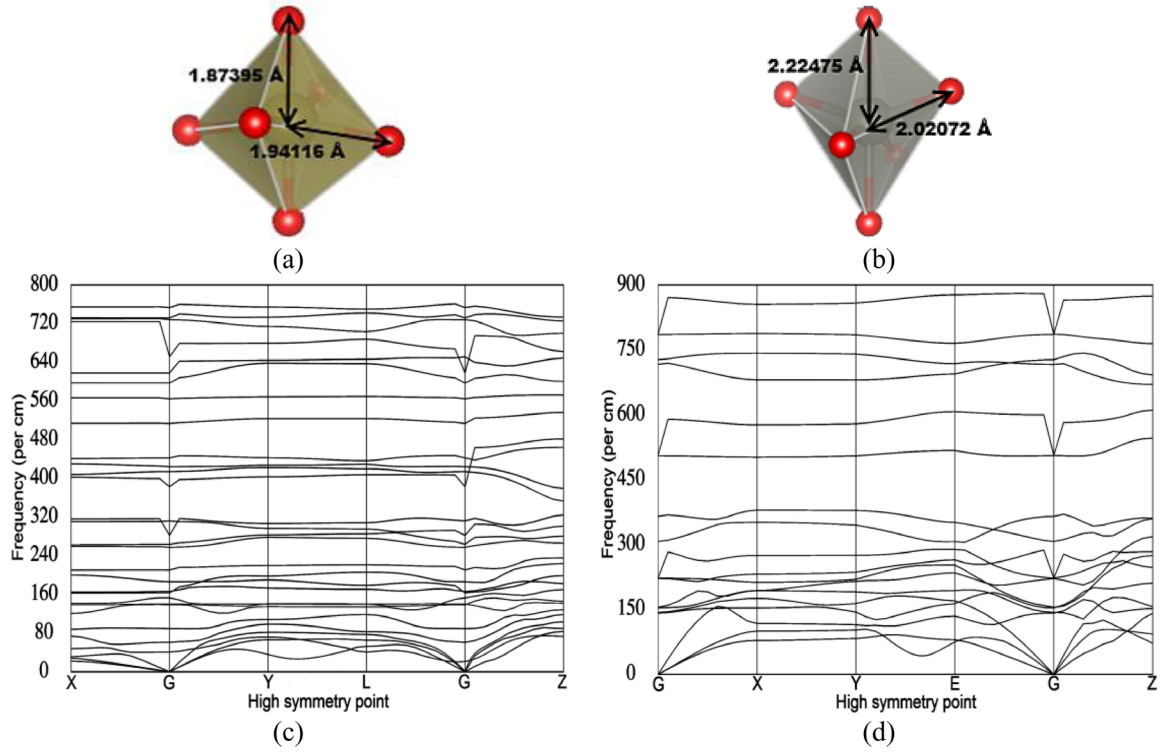


Figure 2. Octahedral framework of (a) $[\text{TeO}_6]$ and (b) $[\text{ZnO}_6]$ units. Phonon dispersion plots of $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 are represented by (c) and (d).

re-relaxing the ionic positions. The electronic relaxation stopped when force between atoms and energy change per atom were below 1×10^{-4} eV \AA^{-1} and 1×10^{-5} eV, respectively. The calculation of the electronic properties was performed using generalized gradient approximation (GGA) of Perdew–Burke–Ernzerhof (PBE) [31] furnished by VASP, in terms of exchange correlation (XC) potentials. Effects of core electrons on valence orbitals were described using projector-augmented wave approach, with plane-wave threshold kinetic energy chosen as 350 eV. Tetrahedron method with Blöchl corrections was utilized in all cases. For the computations of G_0W_0 and linear and nonlinear optical parameters, we employed DFPT [32] as implemented in the ABINIT [33, 34] code, in which norm-conserving pseudopotentials and PBE-GGA [35] exchange-correlation functional were preferred. To avoid having symmetric atomic positions and lattice tolerance inconsistencies that might arise due to different codes, we reoptimized again every study specimen using ABINIT, before executing density functional perturbation theory (DFPT) calculations. Convergence studies indicate that a cutoff of 35 Hartree and $4 \times 4 \times 4$ k -point meshes were enough to produce convergence in energy. Number of unoccupied states equivalent to twice the strict minimum ($0.5 \times$ total valence electron per unit cell), 35 Hartree cut-off energy and denser k -point grids of $12 \times 12 \times 12$ were applied for optical run.

3. Results and discussion

3.1. Structural properties

Towards the end of structural prediction, formation enthalpy (ΔH_{ZnTeO}) of fittest structures was plotted against y , O_2

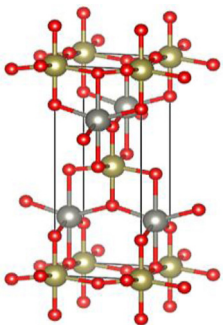
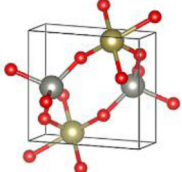
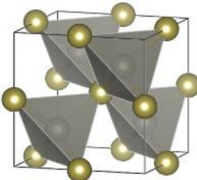
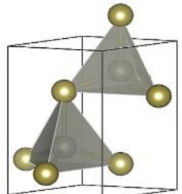
composition. Formation enthalpy of ZnTe-O_2 was evaluated using

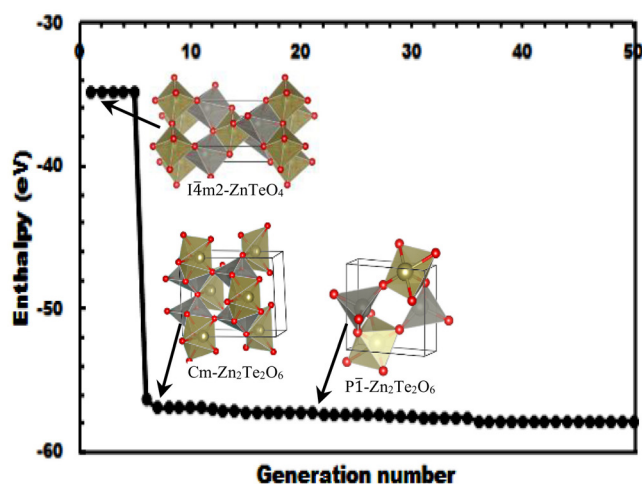
$$\Delta H_{\text{ZnTeO}} = [E_{\text{ZnTeO}} - [xE_{\text{ZnTe}} + yE_{\text{O}}]] / (x + y). \quad (1)$$

A specific structure is classified as energetically stable if it fulfils the following two principles: (1) being a ground state structure with negative formation enthalpy and (2) is close to the lowest lying composition. Figure 1 illustrates the convex hulls of the ZnTeO composites at barometric pressure. It comprises of points that commensurate to phases with the lowest formation enthalpy at exact compositions, in reference to the lowest enthalpy structures of base elements. The perforated line joining all phases with the lowest negative formation enthalpies is termed as a convex hull. As can be seen, global minimum structure or composite with lower Gibbs free energy than any phase accumulation is the triclinic $\text{P}\bar{1}$ $\text{Zn}_2\text{Te}_2\text{O}_6$ compound. Tetragonal $\text{I}4\text{m}2$ ZnTeO_4 was projected as the nearest metastable phase to lowest ground state structure, with approximately 0.05 eV per atom difference between them. Their geometries were added as seeds for further fixed composition calculations, with greater percentage of mutations used as offspring generator. Output produced is consistent with variable composition runs, hence ensuring acquisition of well-relaxed specimens.

Their dynamical stability is confirmed by analyzing the phonon spectrum within whole Brillouin zone in figure 2, where no negative frequencies are observed. The $\text{I}\bar{4}\text{m}2$ structure exhibits an ordered 3D linkage comprised of $[\text{ZnO}_6]$ and $[\text{TeO}_6]$ octahedra. Careful analyses at these units point to incidence of elongated and compressed distortions (see figure 2). End block structure of binary ZnTe was correctly forecasted to adopt a

Table 1. Crystallographic data of fit ZnTeO. Grey, yellow and red denote the Zn, Te and O atoms, respectively. Haynes and Kittel [36, 37] give the experimental values of lattice constants for ZnTe.

Phase	Unit cell	Lattice constants/Å			Atomic Wyckoff positions
		<i>a</i>	<i>b</i>	<i>c</i>	
ZnTeO ₄		3.8435	3.8435	9.9944	Zn 2d (0.000 00, 0.500 00, 0.750 00) O 4e (0.000 00, 0.000 00, 0.187 50) O 4f (0.000 00, 0.500 00, -0.027 40) Te 2a (0.0000, 0.0000, 0.0000)
Zn ₂ Te ₂ O ₆		5.4040	5.2280	5.0360	Zn 2i (0.238 21, -0.171 76, -0.483 40) O 2i (-0.116 84, -0.176 38, -0.374 46) O 2i (-0.401 27, 0.083 23, 0.183 39) O 2i (0.402 38, -0.411 83, 0.240 24) Te 2i (0.301 77, 0.289 32, 0.057 10)
Zn ₂ Te ₂		6.0345 6.1010	6.0345 6.1010	6.0345 6.1010 [36]	Zn 4d (0.750 00, 0.750 00, 0.750 00) Te 4b (0.500 00, 0.500 00, 0.500 00)
Zn ₆ Te ₆		4.2337 4.2700	4.2337 4.2700	6.9670 6.9900 [37]	Zn 2b (0.333 33, 0.666 67, -0.188 91) Te 2b (0.333 33, 0.666 67, 0.184 38)

**Figure 3.** Enthalpy of the best individual versus generation number. Average population size of 30.

zincblende form [36], which transforms to wurtzite layout [37] at slightly higher formation enthalpy. Table 1 summarizes the crystallographic data of aforementioned four phases. Note that the computed physical dimension of Zn₂Te₂ and Zn₆Te₆

unit cells are extremely close to experimental values [36, 37]. Moreover, the calculated powder pattern (see supplementary figure 1 (stacks.iop.org/CM/32/225701/mmedia)) for cubic Zn₂Te₂ also resembles actual XRD analysis [38], thus ensuring the reliability of our applied methodology. Reproduction of experimentally grown ZnTeO₃ and Zn₂Te₃O₈ [39] phases was not detected. Absence of ZnTeO₃ and Zn₂Te₃O₈ is also reported in the work by Nabetani *et al* [40].

The proficiency of the USPEX algorithm in yielding various energetically stable structures is extremely efficient, to which in our adopted input setting only between 30–200 structures (figure 3) were calculated before the metastable ZnTeO₄ and pre-global minimum Zn₂Te₂O₆ phases started appearing. Note that the best Zn₂Te₂O₆ structure underwent a monoclinic to triclinic transition in the 22th generation. Its structure and energy remained constant after the 35th generation.

3.2. Electronic properties

Having examined the primitive cells and energetics of numerous ZnTe-O₂ stoichiometries, we now shift to their electronic properties. Figure 4 shows the total and projected

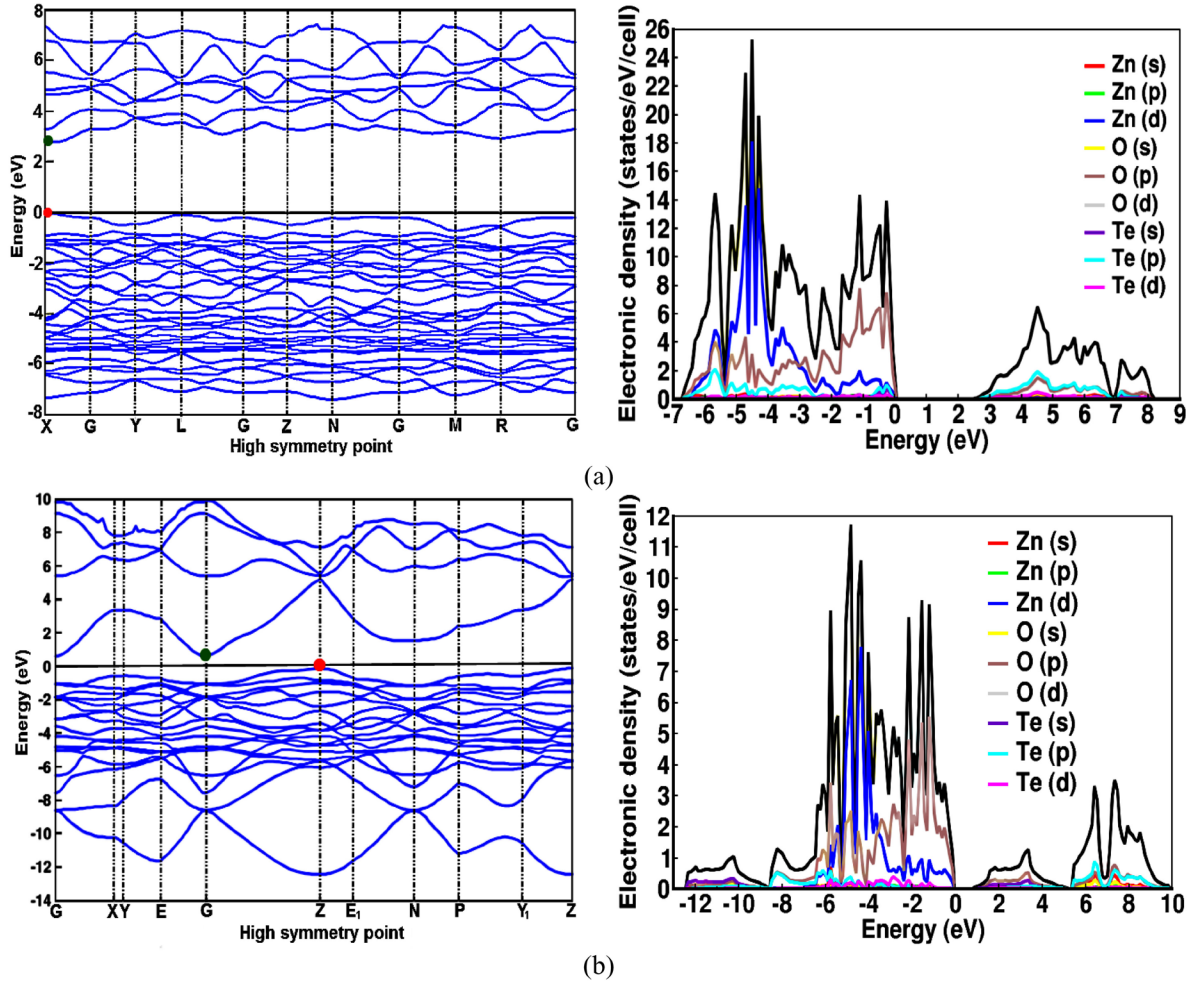


Figure 4. Electronic band structure and projected DOS for (a) $\text{Zn}_2\text{Te}_2\text{O}_6$ and (b) ZnTeO_4 whereby green and red spheres represent the CBM and VBM, correspondingly.

Table 2. Computed KS and G_0W_0 energy gap (eV) for $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 .

	$\text{Zn}_2\text{Te}_2\text{O}_6$		ZnTeO_4
	X	Z	$Z \rightarrow \Gamma$
KS	2.86	5.40	0.74
G_0W_0	4.27	7.25	2.59

density of states (PDOS) for all ground-state composites. It is clearly depicted that their respective PDOS are made up of three pronounced energy regions: (1) an overlapped Zn-*d*/O-*p* and Te-*p*/O-*p* states in the deeper lying valence band (from -2.0 to -7.0 eV for $\text{Zn}_2\text{Te}_2\text{O}_6$, from -3.0 to -9.0 eV for ZnTeO_4), (2) localized Zn-*d* band and (3) a segregated conduction band. The first band is densely occupied by the Zn-*d* orbitals, with contributions from *p* orbitals of O and Te atoms. Hence, the influence of this band to the bonding is extremely strong. Subsequent upper valence band is a result of weaker hybridization from the *d* and *p* states of Zn and O atoms. Addition of O atoms leads to the introduction of localized states, which subsequently go through a reciprocal action with delocalized states of ZnTe host crystal. This interaction is responsible for the splitting of conduction band (located at

~ 7.0 eV and ~ 5.5 eV of figures 4(a) and (b)), a unique feature of most HMAs.

Sharp dissimilarity between their PDOS plots indicate a strong dependence on O composition. In contrast, ZnTeO_4 appears to have a broader but lower peak plot, prompting a wider range of atomic orbitals hybridization. Covalency is enhanced under these circumstances. We solidified this claim by exploiting Lyakhov–Oganov [41] model (<http://han.ess.sunysb.edu/hardness/>), which engages bond strengths and bond topology for hardness calculation. Calculated hardness of $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 are 3.722 GPa and 6.828 GPa, thereby suggesting a positive correlation of covalency with O content. Band diagrams in figure 4 imply a direct and indirect energy gap nature for $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 phases, with maximum-minimal energy states at X and $Z \rightarrow \Gamma$, respectively. The Kohn–Sham (KS) and single-shot G_0W_0 corrected eigenvalues are listed in table 2.

The direct energy difference between conduction band minimum (CBM) and valence band maximum (VBM) appears to widen as O concentration rises. A similar trend has also been reported by Yu *et al* [9] in their work about ZnTeO preparation via insertion of O into ZnTe host alloy. Compounds with such high energy band gap have been projected in certain studies

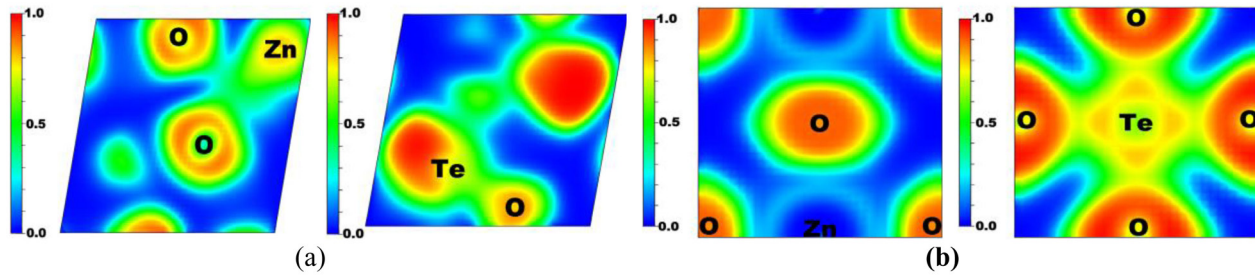


Figure 5. The calculated ELF of (a) triclinic $\text{Zn}_2\text{Te}_2\text{O}_6$ and (b) tetragonal ZnTeO_4 phases.

Table 3. Elastic constants (GPa) for ZnTeO_2 .

Phase	C_{ij}								
	C_{11}	C_{12}	C_{13}	C_{22}	C_{23}	C_{33}	C_{44}	C_{55}	C_{66}
$\text{Zn}_2\text{Te}_2\text{O}_6$	238.72	105.63	107.89	201.95	106.76	107.89	205.27	102.31	87.68
ZnTeO_4	317.79	128.97	116.74	308.80	128.16	116.74	257.80	140.07	126.25

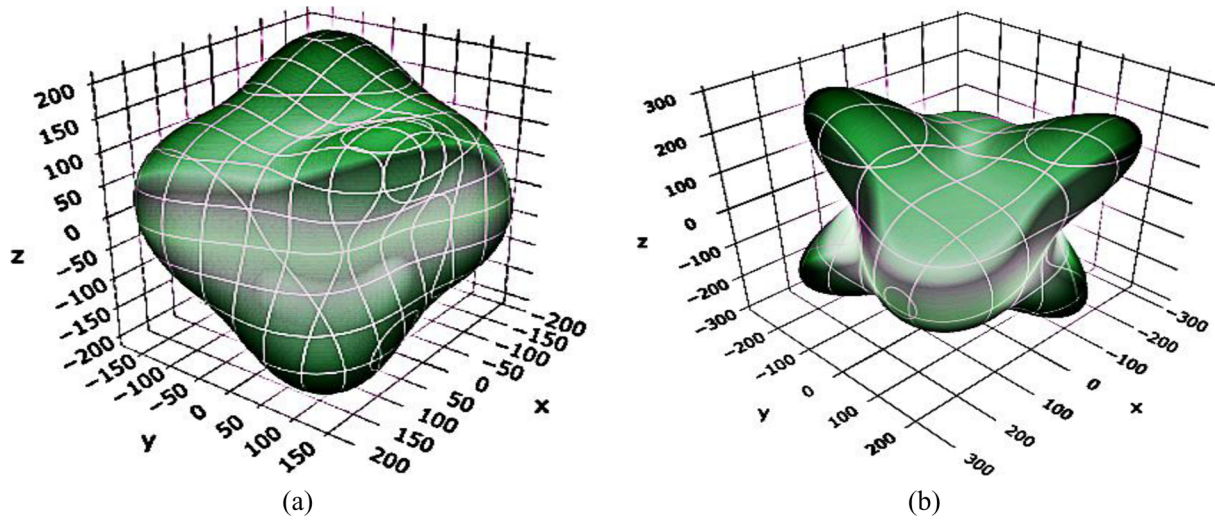


Figure 6. 3D contour plot of Young's modulus.

[42, 43] to exhibit a high level of laser damage threshold. During second harmonic generation (SHG) process, certain amount of SHG signals generated may have energy greater than the band gap of nonlinear optical (NLO) crystal. This results in the production of point defects in the crystal and yields more energy level to the atoms, thus causing optical absorption at definite wavelengths (color center). The increase of absorption at SHG wavelength often leads to crystal damage. As such, materials with large band gap are desired candidates for SHG purpose.

We include herewith (figure 5) the electron localization function (ELF) to comprehend the bonding nature in $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 structures. The estimated ELF of ~ 0.5 for Zn–O and Te–O bonds in triclinic structure corresponds to weak chemical interactions. A different scenario is noted in tetragonal structure, where its higher estimated ELF (~ 0.75) for Te–O bonds signifies an enhanced covalent bonding.

3.3. Optical nonlinearity enhancement via elastic anisotropy

The elastic parameters of a substance are measurable through its reaction to external stress. Contrarywise, the applied stress is mandatory in maintaining specific deformation, and can be used to examine elastic properties that directly quantify the mechanical stability. The computed elastic constants for all predicted stable ZnTeO compounds are stored in table 3. These compounds are deemed having sufficient mechanical stability if the following stability criteria [44, 45] are satisfied:

$$\text{Tetragonal} \rightarrow C_{11} > |C_{12}|, 2C_{13}^2 < C_{33}(C_{11} + C_{12}), C_{44} > 0, C_{66} > 0$$

$$\text{Triclinic} \rightarrow \begin{vmatrix} C_{11} & \cdots & C_{16} \\ \vdots & \ddots & \vdots \\ C_{61} & \cdots & C_{66} \end{vmatrix} > 0.$$

By substituting the elastic parameters in table 3 into above corresponding inequalities, we noticed that all predicted phases satisfy the stability requirements, hence confirming sufficient

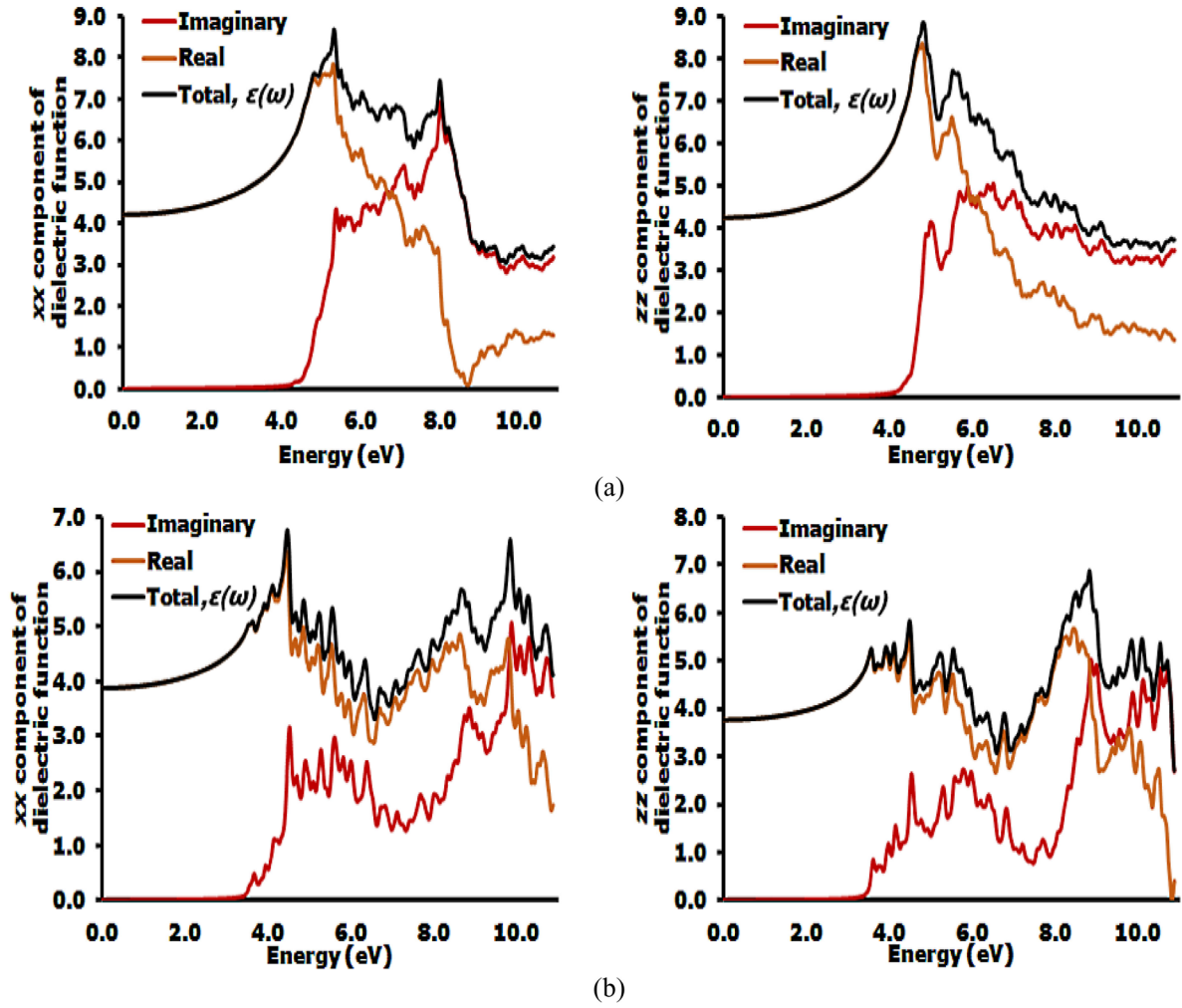


Figure 7. Dielectric functions for (a) $\text{Zn}_2\text{Te}_2\text{O}_6$ and (b) ZnTeO_4 .

mechanical stability in each of them. The values of C_{11} , C_{22} and C_{33} for both phases are quite uneven and as a result events of disproportionate atomic bonding strength along [100], [010] and [001] directions will be apparent. This occurrence is indicative of a high elastic anisotropy degree, a promising condition for the enhancement of optical nonlinearity.

Along the way, we performed the tensors analysis (see supplementary table 1) via online tool ELATE [46]. Their computed Poisson's ratio ν of 0.2808 and 0.2565 for $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 means both phases are of ceramic class [47]. The ν below 0.26 indicates greater covalency [48] in ZnTeO_4 phase, which explains its lower but wider DOS and higher computed hardness. 3D plot of their respective Young's modulus further signifies presence of high anisotropic behavior, particularly ZnTeO_4 phase, as illustrated by the notable spherical shape deformation in figure 6. The universal anisotropy value [49] and percent of anisotropy were also calculated and used to quantify evidence of anisotropy:

$$A^U = 5 \frac{G_{\text{Voigt}}}{G_{\text{Reuss}}} + \frac{B_{\text{Voigt}}}{B_{\text{Reuss}}} - 6 \quad (2)$$

$$A_B = \frac{B_{\text{Voigt}} - B_{\text{Reuss}}}{B_{\text{Voigt}} + B_{\text{Reuss}}} \quad (3)$$

$$A_G = \frac{G_{\text{Voigt}} - G_{\text{Reuss}}}{G_{\text{Voigt}} + G_{\text{Reuss}}}. \quad (4)$$

A^U index is a reliable indicator that furnishes consistent results for elastic anisotropy events in most compounds. Degree of anisotropy is proportional to the deviation of A^U from zero. One can interpret that the A^U value of 0.7498 and 0.9743 hint at high elastic anisotropy in $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 crystalline structures. Both $\text{Zn}_2\text{Te}_2\text{O}_6$ and ZnTeO_4 have lower A^B value (1.11% and 0.75%) in contrast to their A^G (6.78% and 8.75%), which means origin of optical nonlinearity is instigated by anisotropy of shear modulus. This finding suggests that there could be a positive inclination of overall ZnTeO anisotropy nature towards oxygen content. This kind of elastic anisotropy has also been reported to trigger various anisotropic photophysical properties that promote strong optical anisotropy [50].

3.4. Optical properties

Electronic band structure and energy gap play a pivotal role in determining the complex dielectric function $\varepsilon(\omega)$ of semiconducting materials. As such, gap difference between KS and G_0W_0 was included to make the following optical spectrum

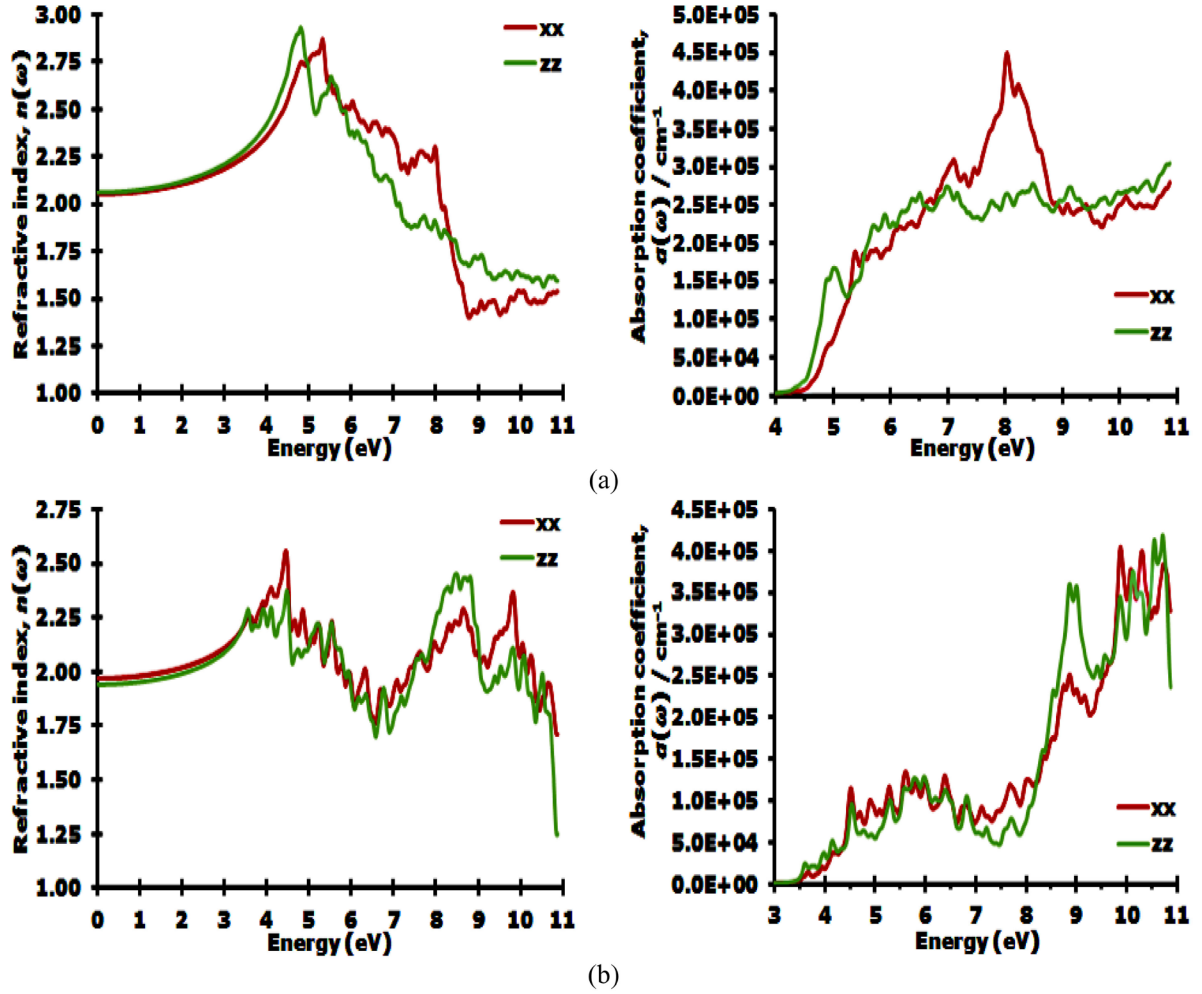


Figure 8. $n(\omega)$ and $\alpha(\omega)$ for (a) Zn₂Te₂O₆ and (b) ZnTeO₄.

computations more realistic. Linear optical response in our work was calculated using random-phase approximation (RPA) [51] within the frame of DFT, describing the interband transitions at absorption region. Omission of indirect intra-band transitions which are associated to free carrier absorption involving phonons scattering is a reasonable approach since both Zn₂Te₂O₆ and ZnTeO₄ phases are ultra-wide band gap materials. For any study specimen, $\varepsilon(\omega) = \varepsilon_1(\omega) + i\varepsilon_2(\omega)$ as a function of incident photon energy is composed of real and imaginary parts. Real component $\varepsilon_1(\omega)$, which is associated with electronic polarizability, can be estimated from its imaginary counterpart $\varepsilon_2(\omega)$ via Kramers–Kronig transformation of imaginary part $\varepsilon_2(\omega)$. We computed the imaginary part of dielectric function from RPA, where complex dielectric function $\varepsilon(\omega) = 1 + 4\pi\chi^{(1)}(-\omega, \omega)$. Summating the transitions from occupied to unoccupied states over the entire Brillouin zone generates $\varepsilon_2(\omega)$. The linear optical susceptibility $\chi^{(1)}$ was scissor corrected by a magnitude corresponding to the difference between the KS band gap and GW corrected band gap, in an effort to include the many-body effects or self-energy effects. The same scissor approximation was also applied to

the calculations of effective second-order susceptibility tensor $\chi^{(2)}$. Equations denoting $\chi^{(1)}$ and $\chi^{(2)}$ are as follows:

$$\chi_{xy}^{(1)}(-\omega, \omega) = \frac{e^2}{\Omega\hbar} \sum_{nmk} f_{nm}(\mathbf{k}) \frac{r_{nm}^a(\mathbf{k}) r_{mn}^b(\mathbf{k})}{\omega_{mn}(\mathbf{k}) - \omega + \frac{\Delta}{\hbar} (\delta_{mc} - \delta_{nc})} \quad (5)$$

$$\chi^{(2)}(-2\omega, \omega, \omega) = \chi_{xyz}^{(2)}(-2\omega, \omega, \omega) + i \frac{\sigma_{xyz}^{(2)}(-2\omega, \omega, \omega)}{2\omega} \quad (6)$$

where Ω , r_{nm} , c and Δ indicate unit cell volume, elements of momentum matrix in k space, conduction states and energy shift, respectively. Subscript letters x , y and z designate the Cartesian components. The first and second terms in (6) represent the pure interband influence and contribution of the susceptibility resulting from the modification of the intraband motion by the polarization energy associated with the interband motion, as described below:

$$\chi_{\text{inter}}^{(2)}(2\omega, \omega, \omega) = \frac{1}{\Omega} \sum_{nmk} W_k \times \left\{ \frac{2r_{nm}^a(r_{ml}^b r_{ln}^c)}{(\omega_{lm} - \omega_{ml})(\omega_{mn} - 2\omega)} - \frac{1}{(\omega_{mn} - \omega)} \left[\frac{r_{lm}^c(r_{mn}^a r_{nl}^b)}{(\omega_{nl} - \omega_{mn})} - \frac{r_{nl}^b(r_{lm}^a r_{mn}^c)}{(\omega_{lm} - \omega_{mn})} \right] \right\} \quad (7)$$

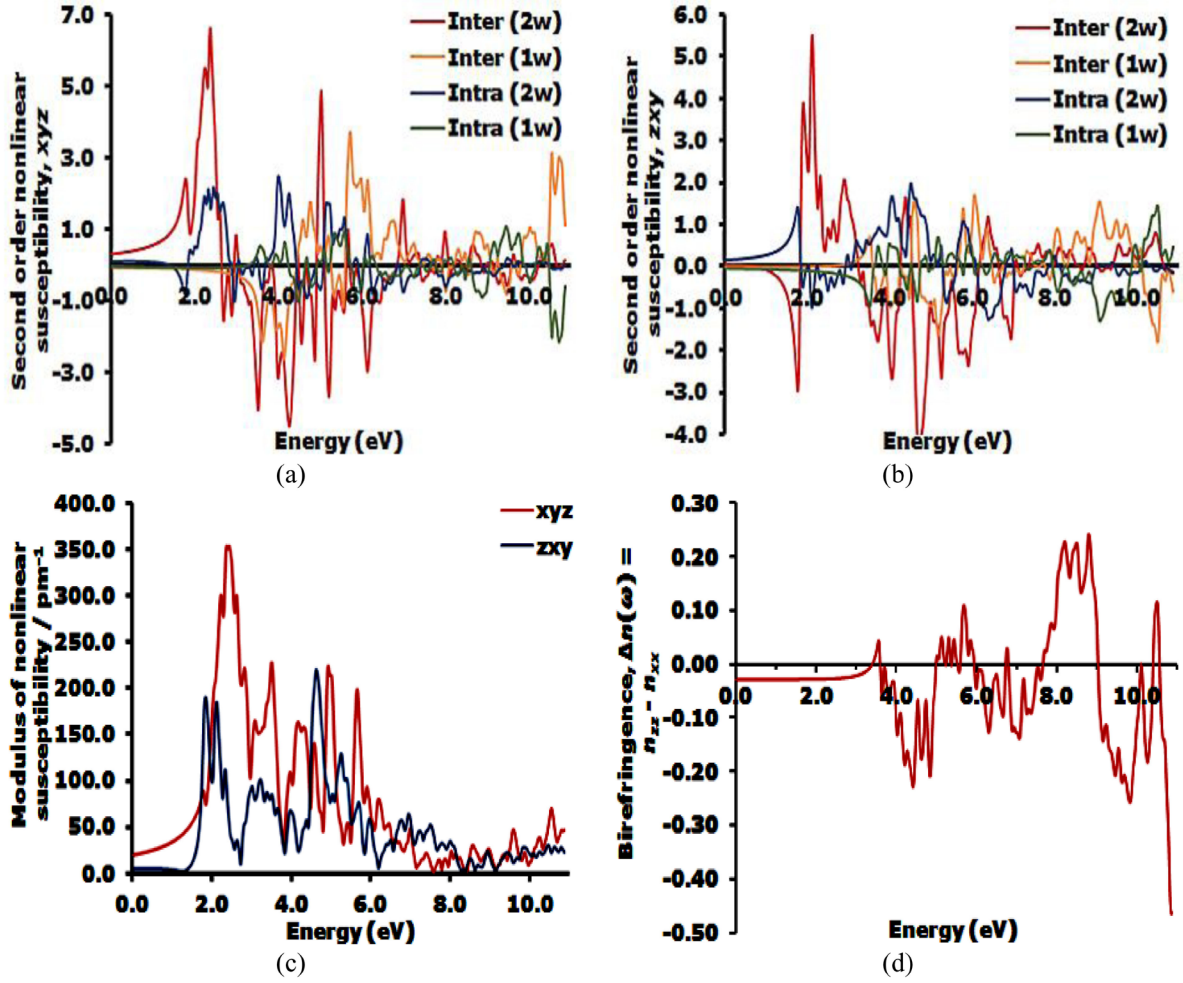


Figure 9. Calculated (a) xyz and (b) zxy components of nonlinear susceptibility, (c) modulus of $\chi^{(2)}$ and (d) birefringence $\Delta n(\omega)$ for ZnTeO₄.

$$\chi_{\text{mod}}^{(2)}(2\omega, \omega, \omega) = \frac{1}{2\Omega} \sum_k W_k \left\{ \sum_{nml} \frac{1}{\omega_{nm}^2(\omega_{mn} - \omega)} \times [\omega_{nl} r_{lm}^a (r_{mn}^b r_{nl}^c) - \omega_{lm} r_{nl}^a (r_{lm}^b r_{mn}^c)] - i \sum_{nm} \frac{r_{nm}^a (r_{mn}^b \Delta_{nm})}{\omega_{nm}^2(\omega_{mn} - \omega)} \right\} \quad (8)$$

where n , m , l and W_k indicate the valence states, conduction states, all states and k -point weightage. The momentum matrix elements r_{nm} are determined by response wavefunction, which is computed via DFPT. The basis of the amended linear susceptibility is originating from an improved Hamiltonian, which has accounted a k independent energy shift, projected on top of the conduction states. A detailed derivation of equation (5) and specific terms in (6) have been presented in the work by Sipe *et al* [52] in their work on independent-particle approximation.

These fundamental properties can then be used to calculate optical spectra such as refractive index $n(\omega)$ and absorption coefficient $\alpha(\omega)$. Detailed descriptions of their calculations are listed in our earlier works [53, 54]. As triclinic and tetragonal crystal classes have different nonzero linear susceptibility, for clarity, we consider the xx and zz elements.

The computed dielectric functions for Zn₂Te₂O₆ and ZnTeO₄ are displayed in figure 7, separately. We notice that ZnTeO₄ with higher O proportion has a more drastic distortion to its $\varepsilon_1(\omega)$ and $\varepsilon_2(\omega)$ curves, as demonstrated by the greater number of dominant peaks formed. From the calculated PDOS in figure 4, we note that the peaks for imaginary part $\varepsilon_2(\omega)$ of ZnTeO₄ below 4 eV are assigned to the $5s \rightarrow 2p$ transitions while the unoccupied $5p$ bands begin after 4 eV. Peaks above 5 eV are attributed to $2s \rightarrow 5p$, $4s \rightarrow 5p$ and $2p \rightarrow 5p$ transitions. In contrast, the peaks for imaginary part $\varepsilon_2(\omega)$ of Zn₂Te₂O₆ are merely ascribed to $2p \rightarrow 5p$ and $4s \rightarrow 2p$ transitions, hence contributing to smaller number of dominant peak structures in Zn₂Te₂O₆ phase. Average static dielectric constant, $\varepsilon(0) = [\varepsilon_{xx}(0) + \varepsilon_{yy}(0) + \varepsilon_{zz}(0)]/3$, decreases from 4.295 to 3.840 as direct band gap increases from 4.265 eV to 7.246 eV. This trend is well described by the Penn model [55], in which $\varepsilon(0) \propto 1/(\text{band gap})^2$.

Figure 8 illustrates the dispersion curves of $n(\omega)$ and $\alpha(\omega)$ for ZnTe-O₂ alloys. Mean zero-limit refractive index, $n(0) = [n_{xx}(0) + n_{yy}(0) + n_{zz}(0)]/3$, declines from 2.055 to 1.960 during triclinic to tetragonal transition. Biaxial and uniaxial birefringence are seen in both crystals, and the distinct refractive indices become prominent in the UV region.

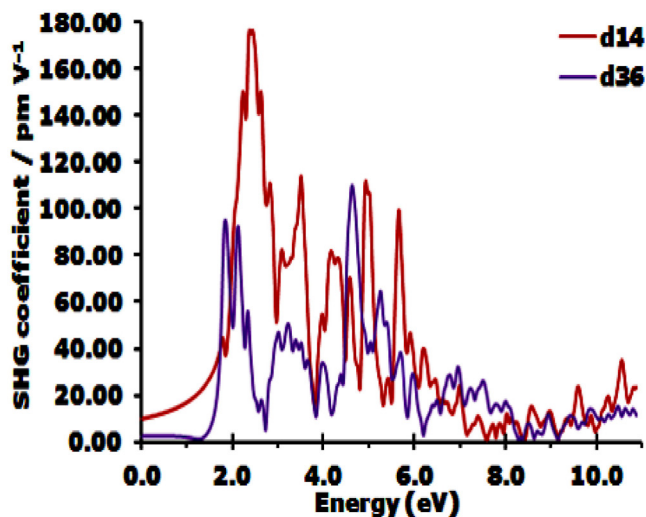


Figure 10. Calculated nonzero SHG coefficients for ZnTeO₄.

It has been a well-known fact that birefringence is one of the catalysts for interesting NLO behaviors [56]. Absorption coefficient of triclinic Zn₂Te₂O₆ starts increasing as light wavelength decreases approaching its band gap, with consistent increase that reaches values of $\alpha > 10^5 \text{ cm}^{-1}$ in UV range. In the case of tetragonal ZnTeO₄, a substantially increased absorption first occurs at its indirect band gap edge, while a steeper increase emerges at the direct band gap point and is extended till deep UV region. The observed high absorption in UV span means these materials are potential candidates for UV trapping system, which uses UV radiation to generate electricity but permits visible light transmission. This wide coverage in UV region also demonstrates the suitability of bulk ZnTeO as host materials for UV-pumped phosphor. Zn₂Te₂O₆ and ZnTeO₄ fall under the crystal class of $\bar{1}$ and $\bar{4}2m$, correspondingly. Due to the influence of Kleinman symmetry [57], nonlinear tensor elements in Zn₂Te₂O₆ cease to exist while ZnTeO₄ has two categories of independent non-vanishing SHG coefficients. These include $d_{14} = d_{25}$ and d_{36} .

Birefringence $\Delta n(\omega)$, imaginary second order susceptibilities and absolute values of χ_{yz} and χ_{xy} components are portrayed in figure 9. Birefringence is an important criterion in determining efficient NLO materials because of its strong correlation with phase-matchability, a phenomenon essential for effective nonlinear interaction. It is noteworthy that ZnTeO₄ has reasonably high birefringence (Δn greatly surpasses 0.08) [58–60] that changes alternately between positive to negative values in the entire UV zone, suggesting its versatility to adopt either quartz and ice or calcite behaviors. The palpable conflicting values of inter/intraband for 2ω and ω resonances gives indication there is considerable enhancement in the anisotropy between $\chi_{yz}^{(2)}$ and $\chi_{xy}^{(2)}$ tensor components when going from the linear optical properties to the nonlinear optical properties. Overall, $\chi_{yz}^{(2)}$ is the dominant tensor component, as indicated by its static limit SHG tensor of 19.68 pm V^{-1} whereas $|\chi_{xy}^{(2)}(0)| = 4.65 \text{ pm V}^{-1}$.

The computed d_{14} and d_{36} ($0.5 \times$ modulus of SHG tensor) in figure 10 range from $9.84\text{--}2.34 \text{ pm V}^{-1}$ and $2.33\text{--}1.13 \text{ pm V}^{-1}$,

spanning the deep infrared till deep UV. These high values are greater than those reported for popular UV optical application material such as borate crystals [61, 62] and could be attributed to the enhanced elastic anisotropy, along with the Jan–Teller distortions of ZnO₆ and TeO₆ octahedral [63].

4. Conclusion

The combination of variable composition screening and first principles techniques have uncovered two energetically, mechanically and dynamically stable ZnTe–O₂ phases under ambient pressure, namely the triclinic $\bar{1}$ -Zn₂Te₂O₆ and tetragonal $\bar{4}2m$ -ZnTeO₄. These phases have relatively wide G_0W_0 corrected band gap that broadens with more oxygen incorporation. Elevation of O is believed to promote covalent bonding via increased hybridization and sequentially improves the hardness. The high and sharp increase of absorption coefficient across UV range propel their suitability as UV absorber, which could possibly be merged with conventional solar panel to maximize light conversion. Interestingly, non-centrosymmetric ZnTeO₄ exhibits many of the desirable aspects in choosing exceptional nonlinear optical materials, as indicated by its high birefringence ($\Delta n \gg 0.08$), significant anisotropy and large SHG coefficients ($d_{14} = 9.84 \text{ pm V}^{-1}$ and $d_{36} = 2.33 \text{ pm V}^{-1}$ at static limit). The present theoretical study is expected to provide a better understanding of material parameters optimization via compositional tuning, required for synthesizing large single ZnTeO compound of optical quality.

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