

Pressure-induced superconductivity in SnSb_2Te_4

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Abstract

Here we firstly report the pressure-induced superconductivity in phase change materials SnSb_2Te_4 . Single crystals of SnSb_2Te_4 were grown using a conventional melting-down method. The resistance under pressure was measured using an originally designed diamond anvil cell with boron-doped diamond electrodes. The temperature dependence of the resistance under different pressures has been measured up to 32.6 GPa. The superconducting transition of SnSb_2Te_4 appeared at 2.1 K (T_c^{onset}) under 8.1 GPa, which was further increased with applied pressure to a maximum onset transition temperature 7.4 K under 32.6 GPa.

Keywords: superconductivity, data-driven science, phase change material

(Some figures may appear in colour only in the online journal)

1. Introduction

Phase change materials (PCMs) are distinguished by their excellent optical and electrical properties and the fast switching between amorphous and crystalline phase and thus are widely used in optics and data storage [1, 2]. Most PCMs are composed of group IV–VI elements, such as GeSb_2Te_4 and GeSb_2Te_5 , and the Fermi level of these compounds at ground states is often located inside the band gap, exhibiting semiconducting properties [3]. Recently, superconducting properties of these binary or ternary PCMs, such as SnTe [4], Sb_2Te_3 [5], GeSb_2Te_4 [6] have been found whenever pressure is applied or carriers are introduced. First-principles calculations have shown that the high-pressure phase ($Pm\bar{3}m$) of SnTe is metallic and has a very flat band near the Fermi level [7], which is favorable to the formation of the Cooper pairs [8]. At high pressure, SnTe can transform into body-center-cubic (BCC) structure with a superconducting critical temperature of ~ 7.5 K [9], which is in accord with the theoretically predicted value of 7.16 K [7]. The superconductivity in Sb_2Te_3 indicates that the crystal structure of the compound remains

stable at lower pressure and a Dirac cone remains stable at pressure of 6.9 GPa [5, 10]. These results provide strong support the occurrence of superconductivity in the lower pressure range of the ambient phase is topological related [10]. GeSb_2Te_4 (GST) can not only undergo phase transition from an amorphous phase (a-GST) to a crystalline (c-GST) by elevated temperatures [11, 12], but also can have a metal-to-insulator transition at elevated pressure [13]. This transition can be explained as a disorder-induced Anderson localization [14]. Recently, pressure-induced superconductivity transition has been reported in GST and shows that this compound undergo a phase transition from an amorphous GST to bcc GST (b-GST) [6, 15]. One interesting point is that superconducting transition disappears at 43.3 GPa at the phase of b-GST, strongly suggesting a second-order quantum phase transition (QPT) [6]. Thus, the high pressure is an effective approach to study the structural properties in this type of compounds without introducing disorders or impurities. The application of pressure has been reported to turn the A_2B_3 -type topological insulator Sb_2Te_3 [5], Bi_2Se_3 [16], and Bi_2Te_3 [17] into superconducting state.

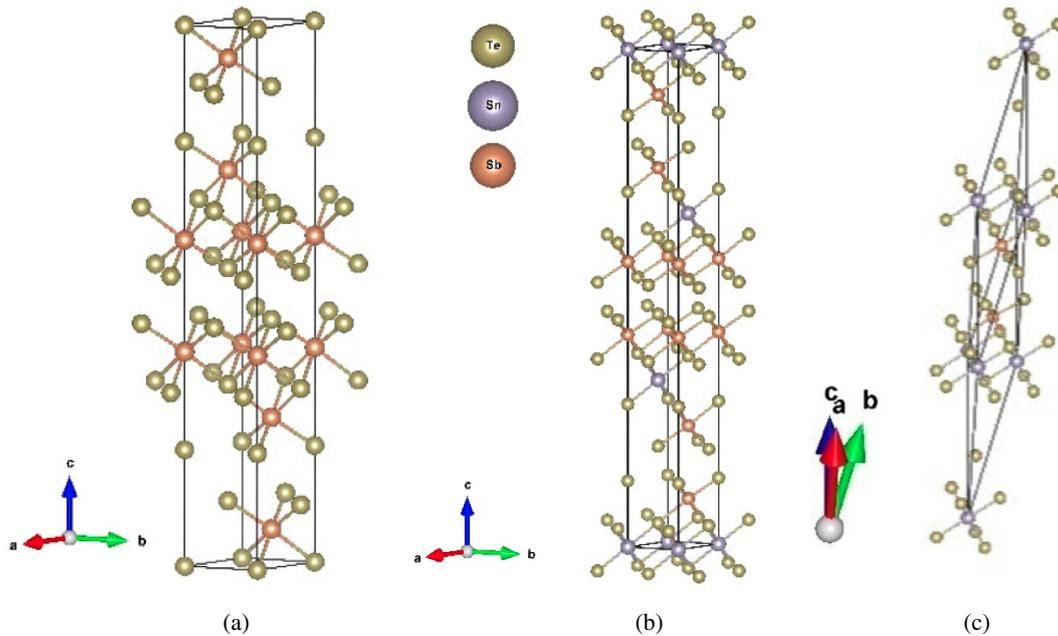


Figure 1. Crystal structures of (a) parent compound of Sb_2Te_3 and (b) SnSb_2Te_4 . (c) Primitive cell of SnSb_2Te_4 . The figures were drawn using VESTA [32]. Brown, green yellow, and gray balls represents for Sb, Te, and Sn atoms, respectively.

Here we focus on the PCM SnSb_2Te_4 , which can be considered as a mixture of SnTe and Sb_2Te_3 , with a similar temperature-dependent phase transition to GST [11, 12, 18] as it can be considered a mixture of SnTe and Sb_2Te_3 . The analysis of both bulk and surface electronic structure indicates that these compounds are also topological related [19] and have a large penetration depth of the topological surface state (TSS) [20]. It is of great interest to check whether this ternary compound would exhibit superconductivity under applied pressure. Herein we studied the pressure-dependent structure and electronic properties on both theoretical and experimental in SnSb_2Te_4 .

2. Method

The structure properties and electronic structure of SnSb_2Te_4 were calculated using the projector augmented wave (PAW) method, as implemented in the Quantum ESPRESSO software package [21, 22]. The generalized gradient approximation (GGA) of Perdew–Burke–Erzerhof (PBE) [23] was used to describe the exchange-correlation function. The inner 4d-electrons in Sn and Sb are treated as valence electrons. A $10 \times 10 \times 10$ k -grid was employed for the k -point sampling in the first Brillouin zone and the kinetic energy cutoffs for the expansion of electronic wave function was set to 74 Ry. The atomic positions and lattice parameters were relaxed using the Broyden–Fletcher–Goldfarb–Shanno (BFGS) algorithm. In the density of states (DOS), a k -mesh size of $20 \times 20 \times 20$ was used. The crystal orbital Hamiltonian population (COHP) [24] and its energy integral (ICOHP) were used for the analysis of chemical bonding, implemented in LOBSTER tool [25].

Single crystals of SnSb_2Te_4 were grown using a conventional melting method. Sn (99.9%, powder), Sb (99.99%,

powder), Te (99.9%, grain), combined in stoichiometric ratios in an evacuated silica tube. Afterward, the tube was put into a furnace and heated up to 1010 K for 10 h. The samples were then slowly cooled to 873 K at a rate of 9.1 K h^{-1} and held for 24 h. Powder x-ray diffraction was conducted by mini flex 600 (Rigaku) with $\text{Cu K}\alpha$ ($\lambda = 1.5406 \text{ \AA}$) radiation. The compositional ratios were investigated by energy-dispersive spectrometry (EDX) using a JSM-6010LA (JEOL). The temperature dependence of resistance under high pressure was measured by boron-doped diamond anvil cells (DAC) [26], implemented in a physical property measurement system (quantum design PPMS). The electrodes at the bottom anvil were fabricated by boron-doped diamond synthesized using a microwave plasma-assisted chemical vapor deposition (MPCVD) technique [27] and were separated by an insulating undoped diamond (UDD) layer. The fabrication process details of this bottom anvil cell are described in [26, 28]. The gasket was a stainless steel with a thickness of $200 \mu\text{m}$. Cubic boron nitride (cBN) powder with ruby manometer was used as a pressure-transmitting medium. The ruby powder was mixed in cBN and was used to measure the applied pressure by the fluorescence from ruby [29] and the Raman spectrum from the culet of top diamond anvil by an inVia Raman microscope (RENISHAW).

3. Results and discussion

Figure 1 shows the crystal structures of SnSb_2Te_4 and its parent compound Sb_2Te_3 . SnSb_2Te_4 has a trigonal structure with space group $R\bar{3}M$ (lattice constants of $a = b = 4.3158 \text{ \AA}$ and $c = 41.6574 \text{ \AA}$) at ambient pressure [19, 30]. The intermediate layers of Sb_2Te_3 and SnSb_2Te_4 are almost identical, and each Sb atom is coordinated with four Te atoms. The structure difference between SnSb_2Te_4 and Sb_2Te_3 is that

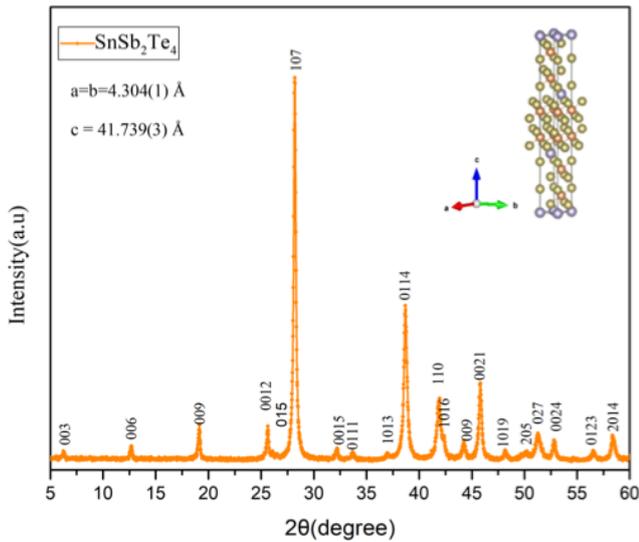


Figure 2. Room temperature x-ray diffraction patterns of SnSb₂Te₄.

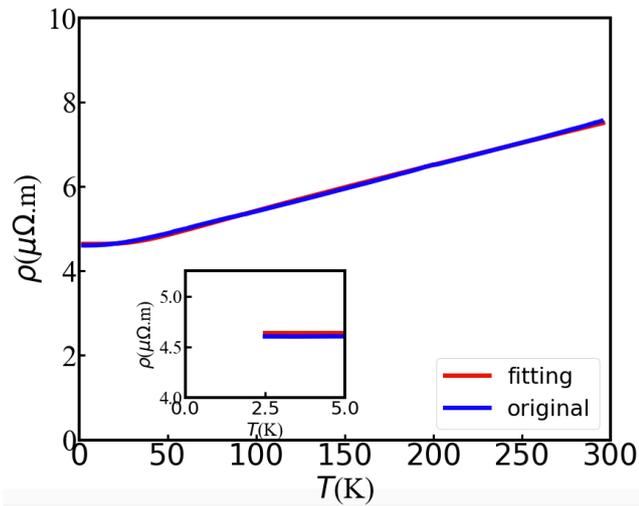


Figure 3. Temperature dependent resistivity of SnSb₂Te₄.

the central Te2 layer in Sb₂Te₃ is replaced by the Te2–Sn–Te2 layer in SnSb₂Te₄. This ternary compound is weakly bonded by van der Waals interaction between Te layers [12, 31]. By a certain heat treatment, the SnSb₂Te₄ also transform into a NaCl-type FCC structure [18]. Figure 2 shows the powder XRD pattern of SnSb₂Te₄. The peaks were corresponding to a $R\bar{3}M(H)$ structure of SnSb₂Te₄, with a lattice constant of $a = b = 4.304(1) \text{ \AA}$ and $c = 41.739(3) \text{ \AA}$. Sn Sb and Te are very uniformly distributed in this material. Using EDS to analyze its composition, the ratio of the material was shown as Sn_{1.01}Sb_{1.98}Te₄, which shows good agreements with nominal composition.

The temperature dependence of the resistivity for SnSb₂Te₄ are shown in figure 3, measured from 2.5 K to 296 K. The resistivity of SnSb₂Te₄ decreases with decreasing temperature which is in good agreement with metallic behavior. In non-magnetic metals, the temperature dependence of resistivity is mainly derived from electron–phonon interactions and can be expressed by Bloch–Gruneissen formula [32].

$$\rho(T) = \rho(0) + \alpha_{\text{el-ph}} \left(\frac{T}{\Theta_R} \right)^5 \int_0^{\frac{\Theta_R}{T}} \frac{x^5}{(e^x - 1)(1 - e^{-x})} dx, \quad (1)$$

where $\rho(0)$ is the residual resistivity due to defect scattering and is independent of temperature, Θ_R is the Debye temperature, and $\alpha_{\text{el-ph}}$ is a constant associated with the electron–phonon coupling constant λ . The red line of figure 3 is the fitted equation from 2.5 K to 296 K and the Debye temperature can be obtained close to 353 K, which is much larger than the Sb₂Te₃ ($\Theta_R = 200 \text{ K}$) [33] and SnTe ($\Theta_R = 165 \text{ K}$) [34]. It is known from the BCS theory that the Debye temperature is positively correlated to the critical temperature T_c [35]. Therefore, This will also convince us that SnSb₂Te₄ can exhibit superconducting properties under pressurized conditions.

Figure 4 shows the calculated band structure, DOS, and COHP of SnSb₂Te₄ at ambient pressure. We also present the band structure of SnSb₂Te₄ (figure 4(d)) under high pressure of 10 GPa for a comparison. We can see that this compound is a semiconductor with a narrow band-gap (0.26 eV obtained by the present GGA-PBE calculations) at the Z point. The energy bands near the top of valence bands and the bottom of conduction bands are both flat, and thus the valence and conduction band edges have high DOS. The orbital-decomposed DOS reveals that the conduction band of SnSb₂Te₄ is mainly composed of Sn-5p, Sb-5p and Te-5p states, and there is also an additional contribution from Te-5s states. The valence bands from -5 eV to -1 eV are mainly contributed by the hybridization between the Sn-(and Sb-) 5p and Te-5p orbitals, while the bands from -1 eV to the valence band maximum (VBM) are contributed by the hybridization between the Sn-(and Sb-) 5s states between the Te 5p orbitals. Figure 4(c) shows the calculated $-p\text{COHP}$ for the nearest-neighboring atom pairs of Sn–Te and Sb–Te, in which the positive (negative) of $-p\text{COHP}$ values correspond to the bonding (anti-bonding) characteristics. Therefore, we can see that the Sn-5p (and Sb-5p) and Te-5p orbitals form the bonding states in the energy range from -5 eV to -1 eV , while the Sn-5s (Sb-5s) and Te-5p orbitals form the anti-bonding states in the energy range from -1 eV to the VBM. The anti-bonding states, which come from the hybridization between the Sn-5p (Sb-5p) and Te-5p, appear in the vicinity of conduction band bottom (CBM).

Compared with the Sb–Te bond length in Sb₂Te₃, the Sn–Te and Sb–Te bond lengths in SnSb₂Te₄ have negligible change (0.02 Å at most). The characteristics of chemical bonding between Sb–Te in two compounds are very similar. It is noted that the energy levels of Sn 5s (5p) are shallower than those of Sb. Therefore, SnSb₂Te₄ has a smaller energy gap than Sb₂Te₃. Figure 4(d) shows the band structure of SnSb₂Te₄ under pressure of 10 GPa. It can be seen that the Fermi level crosses the top valence bands, indicating the pressure-induced a decrease of the band gap of SnSb₂Te₄ and a transition to metallic behavior. As mentioned above, the anti-bonding characteristic appears both in the vicinity of VBM and CBM of SnSb₂Te₄. If the crystal structure does not transform into a different phase under high pressure, the applied pressure will usually lead to a decrease of bond lengths between atoms. For SnSb₂Te₄, the

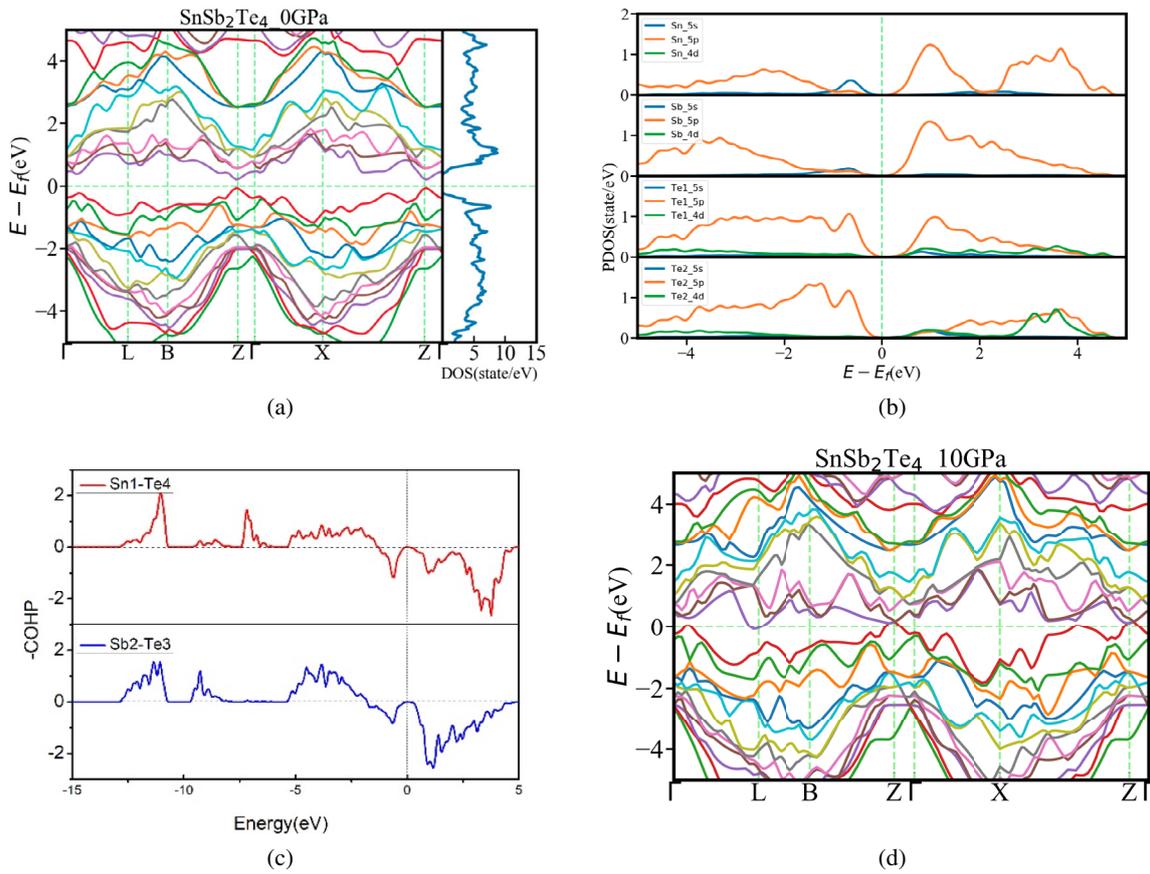


Figure 4. (a) Band structure of SnSb_2Te_4 and electron DOS of SnSb_2Te_4 (b) projected density of state (PDOS) (c) the calculated crystal orbital Hamilton population for the SnSb_2Te_4 at zero pressure (d) the comparative band structure at 10 GPa.

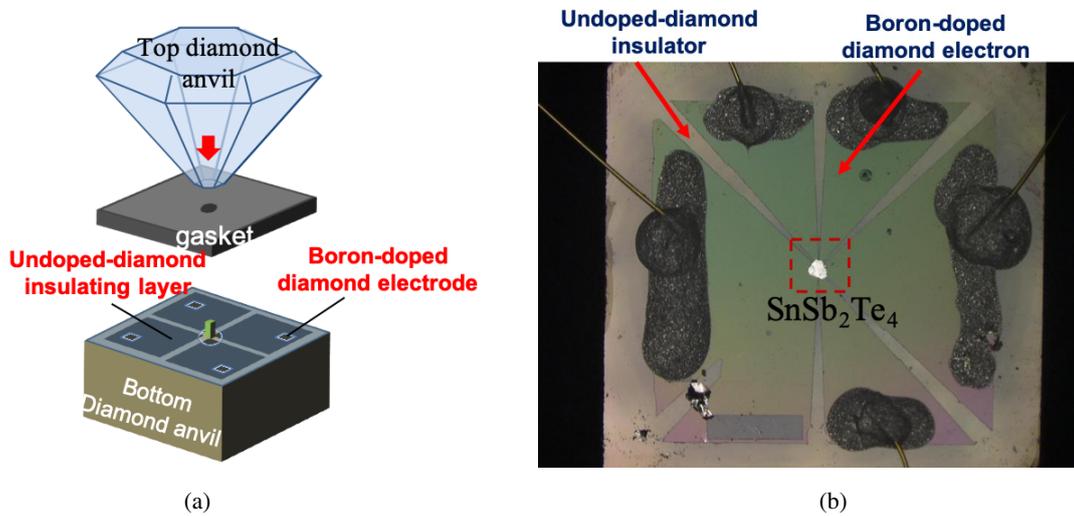


Figure 5. (a) Boron-doped diamond anvil cell, (b) the image of sample in the bottom diamond anvil.

applied pressure results in the shrinking of Sn–Te and Sb–Te bond lengths and thus the anti-bonding states near the VBM and CBM would both shift toward high energy. This might be the reason why the band gap of SnSb_2Te_4 decreases as the applied high pressure. The similar trend was also found in our previous studies for SnBi_2Se_4 [36], PbBi_2Te_4 [37], and AgIn_5Se_8 [37].

Figure 5 shows measurement system of temperature-dependence of resistance using the originally designed

diamond-anvil cell [26]. The sample voltage is detected by heavily boron-doped metallic diamond (BDD) electrodes on the bottom anvil. In measurement of SnSb_2Te_4 , a five-probe design of electrodes was used as shown in figure 5(a). Figure 5(b) shows an image of the sample space of our DAC. The sample was placed at the center of the bottom anvil and measured by a standard four-probe method. Prior to applying pressure, the crystal was cleaved with a scotch tape in order to remove the oxidized surface.

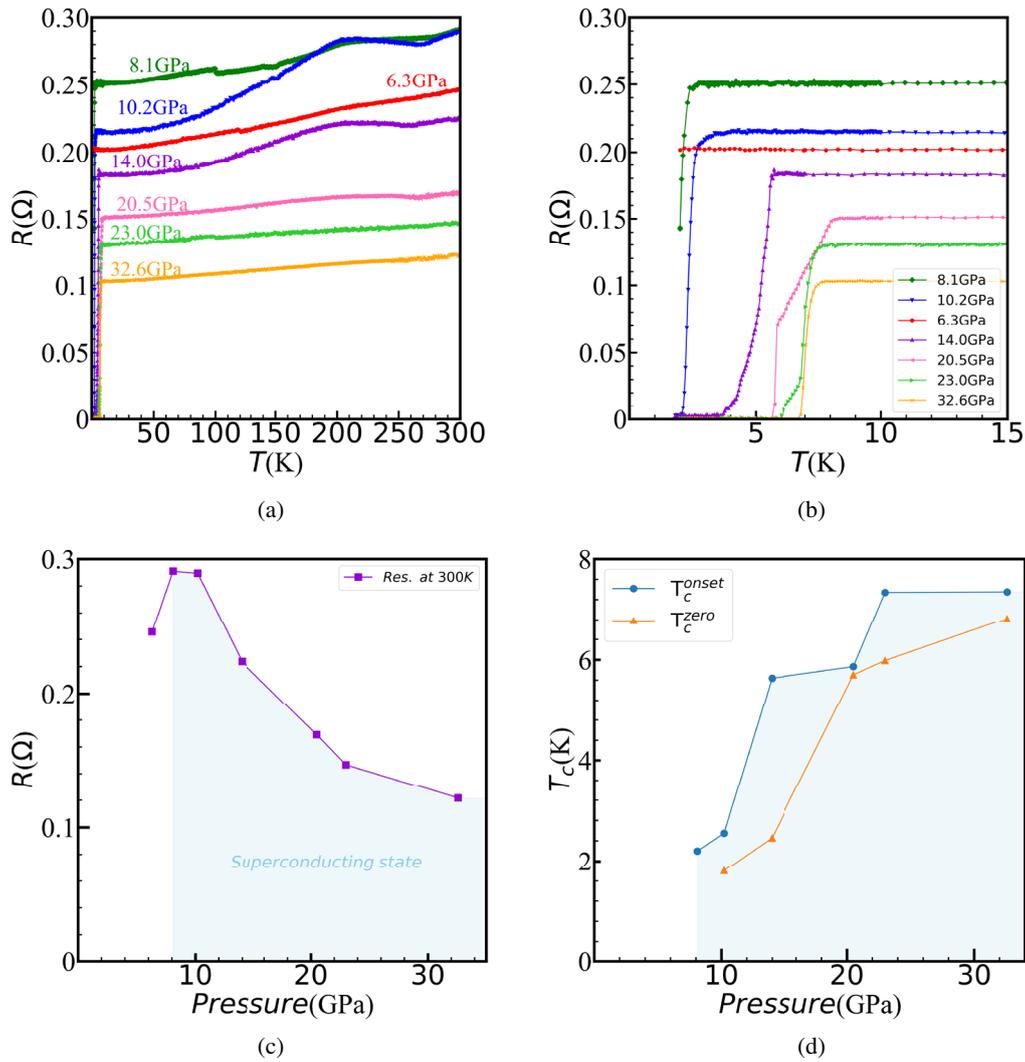


Figure 6. (a) temperature dependence of resistance from 2 K to 300 K (b) temperature dependence of resistance from 2 K to 15 K (c) pressure dependence of resistance from 6.3 GPa to 32.6 GPa (d) superconducting phase diagram.

Figure 6(a) shows the transport measurements at low temperature at pressures between 6.3 GPa to 32.6 GPa. Under these pressures, the measured temperature-dependent resistance of SnSb_2Te_4 reveals a significant metallic behavior for the temperature above 10 K. It is also apparent that superconductivity begins to occur at 8.1 GPa with $T_c^{\text{onset}} \sim 2$ K and the zero resistivity was observed at 10.2 GPa with $T_c^{\text{zero}} \sim 2.2$ K. At a further increase of pressure, in the range between 10.2 and 23 GPa, the superconducting transition temperature increases with increasing pressure. Figure 6(c) shows pressure-dependent electrical resistance at 300 K. From 6.3 GPa to 10.2 GPa, the resistance appears to rise first and then decrease. From our band structures of SnSb_2Te_4 under high pressure, we found that the applied pressure at 10 GPa lead to the closing of band gap of SnSb_2Te_4 , i.e. a semiconductor-to-metal transition. At 10.2 GPa, a resistance kink was observed in the $R-T$ curve. The same phenomenon also occurs in charge-density-wave (CDW) materials such as VSe_2 [38] and NbTe_4 [39]. In NbTe_4 , CDW is gradually suppressed to disappear as pressure increases and superconductivity emerges. In VSe_2 , CDW increases with increasing pressure. Above 12 GPa, CDW state

is suppressed, and superconducting phase begins to appear at 15.2 GPa. These two materials have certain similarities in the crystal structure with SnSb_2Te_4 . Such a kink in the pressure-dependent $R(300\text{ K})$ around 10–12 GPa may be due to the pressure-induced phase transition.

Figure 6(d) shows the superconducting phase diagram of the SnSb_2Te_4 single crystal. From 14.0 GPa to 23.0 GPa, there are some differences between the T_c^{zero} and T_c^{onset} values. It can be seen from figure 6(b) that there are two superconducting transition temperatures at these points, which are likely to have structure phase transition in these pressure range. As the pressure increased, this compound was transferred into a more stable structure phase, resulting in a unique sharp superconductor transition. In the parent compound Sb_2Te_3 , from 22.3 GPa to 26.9 GPa, some $R-T$ curves similar to SnSb_2Te_4 were also observed. And it can be seen from its phase diagram that there are multiple phases of Sb_2Te_3 under these pressures [5]. Similarly, the same phenomenon was observed in the GeSb_2Te_4 [6]. Therefore, in SnSb_2Te_4 , there are likely to be multiple phases from 14.0 GPa. According the context of Bardeen–Cooper–Schrieffer (BCS) superconductivity

theory [40], $\kappa T_c = \hbar\omega_{\text{ph}}e^{-1/\lambda N(E_F)}$, where κ is the Boltzmann constant, $\hbar\omega_{\text{ph}}$ is a phonon energy, λ is the electron–phonon coupling constant, and $N(E_F)$ is the DOS at the Fermi level. Therefore, increasing the DOSs at the Fermi level contributes to the transition from a normal conductor to a superconductor. We have found that in the cases of semiconductor compounds at ambient pressure with anti-bonding states near valence and conduction band edges, the energy gap tends to decrease with increasing pressure when the applied pressure is not large. Based on our previous work [36, 37], the compounds with flat band near the Fermi level and an anti-bonding state in bonding region were very likely to show the pressure-induced superconductivity. Pressure-induced superconductivity in SnSb_2Te_4 provided a strong support to this hint.

4. Conclusion

We studied the band structure information of SnSb_2Te_4 via first-principles calculation. We found that SnSb_2Te_4 had anti-bonding state at the valence and conduction band edges. It should be noticed that many pressure-induced superconductors in our previous papers share similar characteristics. The SnSb_2Te_4 undergoes a superconducting transition with $T_c^{\text{onset}} \sim 2$ K when a pressure of 8.1 GPa is applied. With further increase of pressure, the systems reach a maximum of $T_c^{\text{onset}} \sim 7.4$ K, around 32.6 GPa. With increase of the pressure up to 32.6 GPa, T_c enhances continuously without saturation. Based on the existing R – T data, we have also made some discussions to provide some reference for future researchers.

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